

# TAVOLA PERIODICA DEGLI ELEMENTI

<http://www.ktf-split.hr/periodni/it/>

GRUPPO	TAVOLA PERIODICA DEGLI ELEMENTI																18		
PERIODO	1	2	GRUPPO IUPAC										13	14	15	16	17	VIIIA	
	1A	2A	3A	4A	5A	6A	7A	8A	9A	10A	11A	12A	IIIA	IVA	VA	VIA	VIIA	VIIIA	
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	1.0079 <b>H</b> IDROGENO																	4.0026 <b>He</b> ELIO	
2	6.941 <b>Li</b> LITIO	9.0122 <b>Be</b> BERILLIO			10.811 <b>B</b> BORO								10.811 <b>B</b> BORO	12.011 <b>C</b> CARBONIO	14.007 <b>N</b> AZOTO	15.999 <b>O</b> OSSIGENO	18.998 <b>F</b> FLUORO	20.180 <b>Ne</b> NEO	
3	22.990 <b>Na</b> SODIO	24.305 <b>Mg</b> MAGNESIO											26.982 <b>Al</b> ALLUMINIO	28.086 <b>Si</b> SILICIO	30.974 <b>P</b> FOSFORO	32.065 <b>S</b> SOLFO	35.453 <b>Cl</b> CLORO	39.948 <b>Ar</b> ARGO	
4	39.098 <b>K</b> POTASSIO	40.078 <b>Ca</b> CALCIO											65.39 <b>Zn</b> ZINCO	72.64 <b>Ge</b> GERMANIO	74.922 <b>As</b> ARSENICO	78.96 <b>Se</b> SELENIO	79.904 <b>Br</b> BROMO	83.80 <b>Kr</b> CRIPTO	
5	85.468 <b>Rb</b> RUBIDIO	87.62 <b>Sr</b> STRONZIO											112.41 <b>Cd</b> CADMIO	118.71 <b>Sn</b> STAGNO	121.76 <b>Sb</b> ANTIMONIO	127.60 <b>Te</b> TELLURIO	126.90 <b>I</b> IODIO	131.29 <b>Xe</b> XENO	
6	132.91 <b>Cs</b> CESIO	137.33 <b>Ba</b> BARIO	57-71 <b>La-Lu</b> Lantanidi										200.59 <b>Hg</b> MERCURIO	207.2 <b>Pb</b> PIOMBO	208.98 <b>Bi</b> BISMUTO	(209) <b>Po</b> POLONIO	(210) <b>At</b> ASTATO	(222) <b>Rn</b> RADON	
7	(223) <b>Fr</b> FRANCIO	(226) <b>Ra</b> RADIO	89-103 <b>Ac-Lr</b> Attinidi																

■ Metalli    ■ Semimetali    ■ Non metalli  
1 Metalli alcalini    16 Calcogeni  
2 Metalli alcalino terrosi    17 Alogeni  
3 Metalli di transizione    18 Gas nobili  
4 Lantanidi    5 Attinidi  
**Ne** - gas    **Fe** - solido  
**Ga** - liquido    **Tc** - artificiali

## LANTANIDI

57 138.91 <b>La</b> LANTANIO	58 140.12 <b>Ce</b> CERIO	59 140.91 <b>Pr</b> PRASEODIMIO	60 144.24 <b>Nd</b> NEODIMIO	61 (145) <b>Pm</b> PROMETIO	62 150.36 <b>Sm</b> SAMARIO	63 151.96 <b>Eu</b> EUROPIO	64 157.25 <b>Gd</b> GADOLINIO	65 158.93 <b>Tb</b> TERBIO	66 162.50 <b>Dy</b> DISPROSIO	67 164.93 <b>Ho</b> OLMIO	68 167.26 <b>Er</b> ERBIO	69 168.93 <b>Tm</b> TULIO	70 173.04 <b>Yb</b> ITTERBIO	71 174.97 <b>Lu</b> LUTEZIO
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## ATTINIDI

89 (227) <b>Ac</b> ATTINIO	90 232.04 <b>Th</b> TORIO	91 231.04 <b>Pa</b> PROTOATTINIO	92 238.03 <b>U</b> URANIO	93 (237) <b>Np</b> NETTUNIO	94 (244) <b>Pu</b> PLUTONIO	95 (243) <b>Am</b> AMERICIO	96 (247) <b>Cm</b> CURIO	97 (247) <b>Bk</b> BERKELIO	98 (251) <b>Cf</b> CALIFORNIO	99 (252) <b>Es</b> EINSTEINIO	100 (257) <b>Fm</b> FERMIO	101 (258) <b>Md</b> MENDELEVIO	102 (259) <b>No</b> NOBELIO	103 (262) <b>Lr</b> LAWRENTIO
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(1) Pure Appl. Chem., 73, No. 4, 667-683 (2001)  
Relative atomic mass is shown with five significant figures. For elements that have no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element.

However three such elements (Th, Pa, and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.